

Fig. 4—Hot ductility test results for 56072 (304)

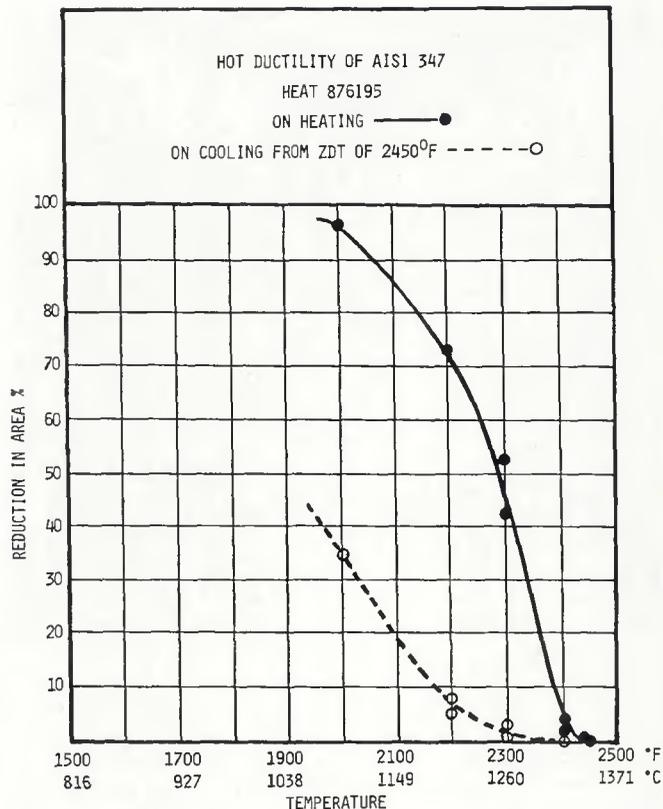


Fig. 5—Hot ductility test results for 876195 (AISI 347)

Metallographic examination via optical and scanning electron microscopy was conducted on the fractured hot ductility samples which were tested in the argon atmosphere. The evidence of liquid films which existed at the instant of fracture was easily detectable on the clean fracture surfaces.

A typical fracture surface from AISI 303 (Heat 0A1090) tested on-heating at the ZDT of 2500°F (1371°C) is shown in Fig. 6A at 20X. The higher magnification fractograph in Fig. 6B shows evidence of a

liquid film which existed at the instant of fracture. However, the microfracture mode is difficult to discern due to the extensive prior liquid film on the fracture surface. The intergranular nature of the fracture is clearly seen in the longitudinal section of the hot ductility sample shown optically at 300X in Fig. 7A. Secondary cracking in regions near to the fracture surface can also be seen. In Fig. 7B, evidence of grain boundary liquation that occurred during the thermal cycle and the subsequent propagation of cracks

along the liquated grain boundary are clearly delineated. Thus, intergranular fracture at the ZDT is related to grain boundary liquation.

#### Varestraint Hot Cracking Tests

Multipass Varestraint tests were conducted to examine the fusion zone, weld metal HAZ and base metal HAZ cracking propensity as described earlier. The total crack length in each of the weld regions was measured and was plotted versus



Fig. 6—Typical fracture surface in heat 0A1090 (303) tested on-heating to a ZDT of 2500°F. SEM. A—20X; B—5000X



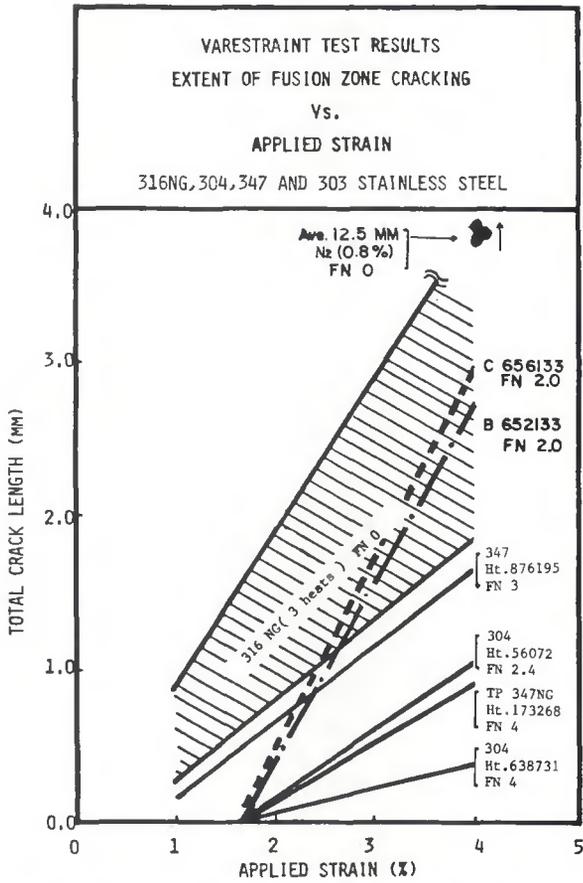


Fig. 8—Extent of fusion zone cracking vs. applied strain in 303 (B652133 and C656133), 304, 347 and 316NG steels

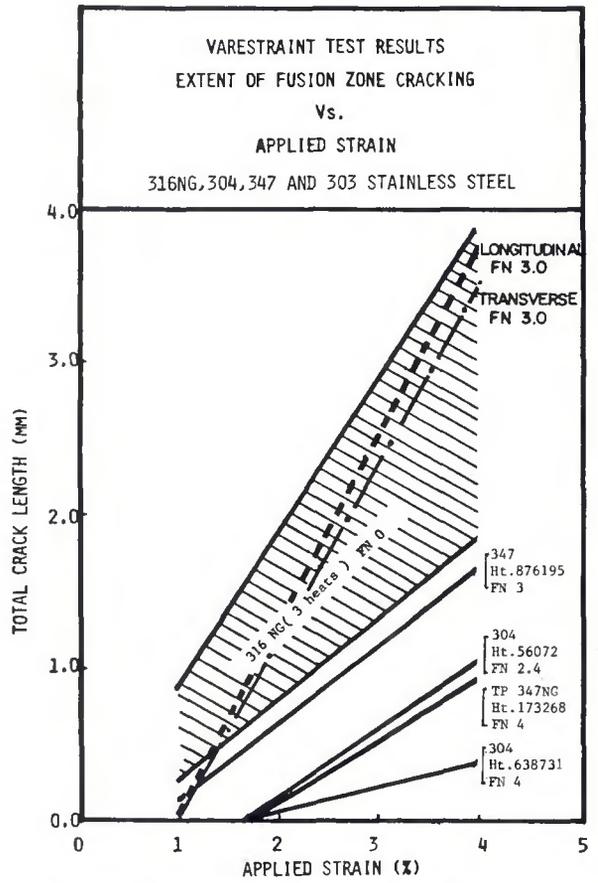


Fig. 9—Extent of fusion zone cracking vs. applied strain in 303 (656471), 304, 347 and 316NG steels

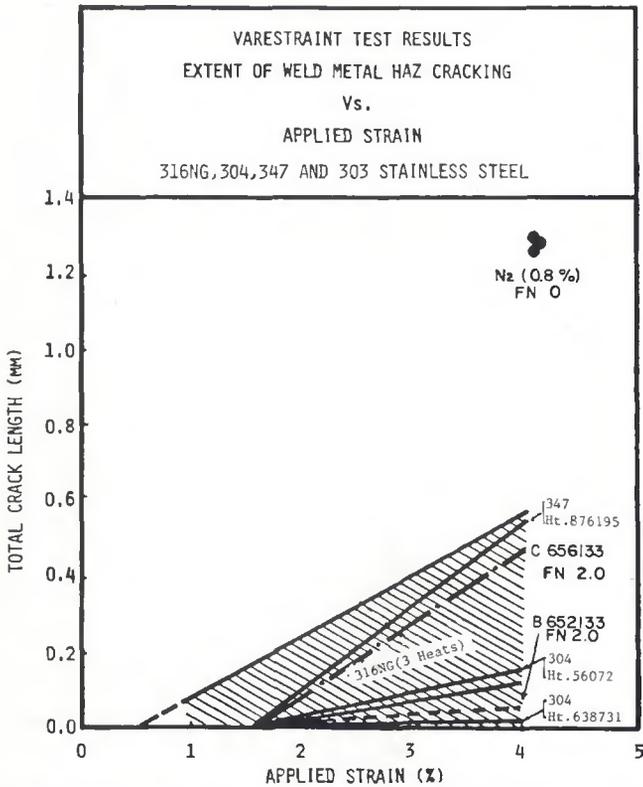


Fig. 10—Extent of weld metal HAZ cracking vs. applied strain in 303 (B652133 and C656133), 304, 347 and 316NG steels

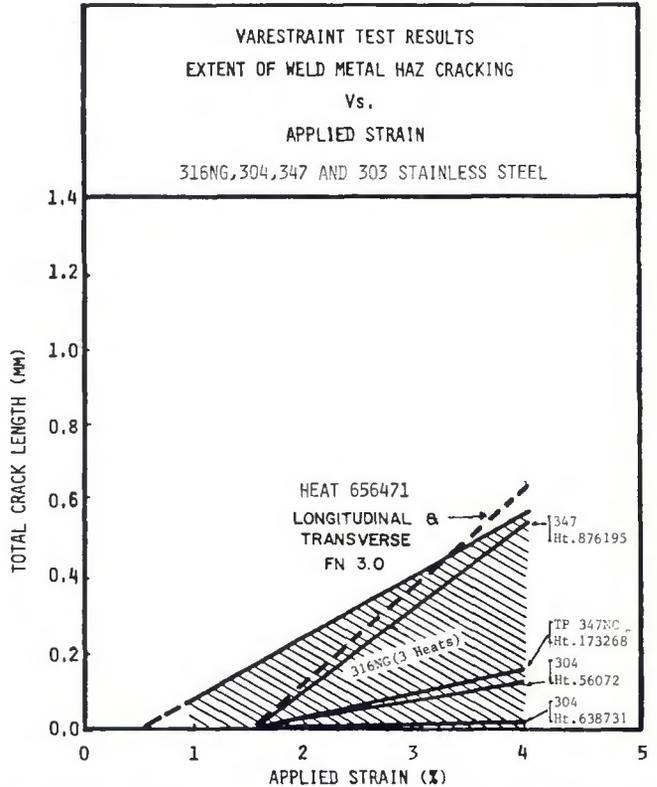


Fig. 11—Extent of weld metal HAZ cracking vs. applied strain in 303 (656471), 304, 347 and 316NG steels







complex sulfides in the fusion zone of the 0.2 ~ 0.22% S-containing Type 304 and 310 stainless steels. Their EDAX analysis showed that sulfides were composed of S, Mn, Cr, Fe and Ni. The exact compositional nature of sulfide in the weld metal is not known, and further studies with TEM and STEM are required to clearly define the type.

#### Solidification Behavior

To observe the high-temperature solidification microstructure, samples from 303 Heat B652133 were rapidly quenched in ice water during welding (with and without the nitrogen addition to the Ar shielding). The quenched weld metal enables the primary solidification modes to be defined and permits comparison of the high-temperature microstructure with the room-temperature microstructure (ferrite content and morphology). As has been indicated, the samples welded with the nitrogen-containing shielding gas were shown to have a higher susceptibility to hot cracking in the fusion zone than those welded with 100% Ar shielding. The higher fusion zone hot cracking susceptibility was related to the low ferrite content (FN  $\approx$  0 with N<sub>2</sub>) at room temperature.

The instantaneous microstructures at the solid-liquid interface regions of 303 Heat B652133, under Ar + 0.8 vol-% N<sub>2</sub> and Ar shielding conditions, are shown in Figs. 19A and B, respectively. In the etched condition (chrome-acetic acid solution), the  $\alpha$ -ferrite appears dark and the  $\gamma$ -phase has a light appearance. In the sample welded with 100% Ar shielding (Fig. 19B), the darkly etched dendrite stems have advanced to the solidification interface, revealing that the sample has a primary ferrite solidification mode. However, in the sample welded with the small



Fig. 19—Instantaneous microstructure of the solid-liquid interface region. A—Ar + 0.8 vol-% N<sub>2</sub>; B—100% Ar. 100X. OLM

addition of nitrogen (0.8 vol-% N<sub>2</sub>) to the shielding gas (Fig. 19A), the well-defined lightly etched dendrites are close to the solidification front, indicating that the cellular dendrites solidified as austenite in the early stages of freezing. Thus, with the small addition of nitrogen, which is a strong austenite stabilizing element, the solidification changed from a primary ferritic to a primary austenitic mode. The darkly etched regions between the primary and secondary dendrite arms, in regions next to the S-L interface, are not  $\delta$ -ferrite, but were liquid at the moment of the sudden quenching in ice water and have retained their own segregation pattern upon rapid cooling. The existence of liquid at the moment of quenching is clearly seen at a higher magnification in Fig. 20A. Figure 20A further shows that during quenching, a crack formed and propagated along a grain boundary.

The evidence for liquid between dendrite arms/stems at the moment of

quenching also can be noted in the sample tested with 100% Ar shielding, as shown in Fig. 20B. The prior liquid drops are indicated by arrows. Figure 20B reveals that the region near the solidification front (A) has a ferrite content greater than 60% at the high temperature at the moment of quenching. In regions behind the solidification front (B and C), which were at lower temperatures than region A at the moment of quenching, the ferrite level is less, due to the partial transformation of ferrite to austenite during cooling below the solidification temperature.

The results of this preliminary solidification study clearly indicate that the fusion zone hot cracking resistance of the 303 alloy is directly related to the solidification modes. Alloy compositions which solidify with a primary ferritic solidification mode have a higher hot cracking resistance than those which solidified with a primary austenitic solidification mode.

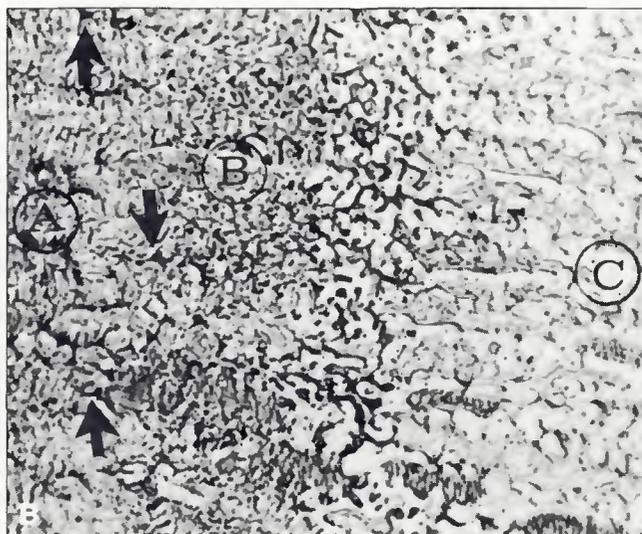
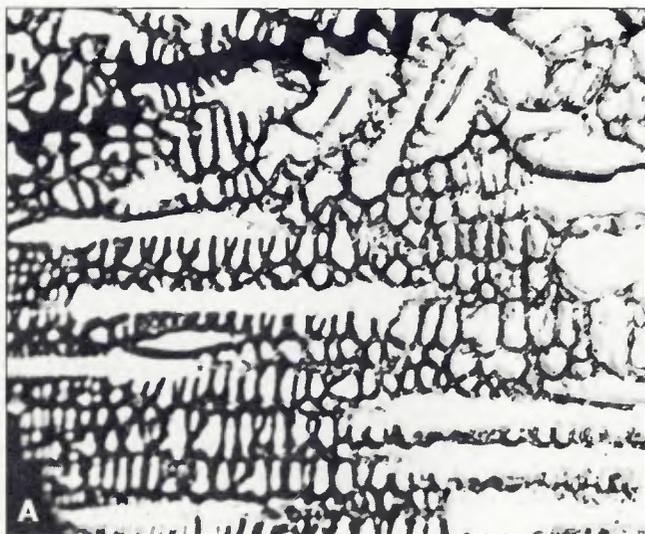


Fig. 20—Higher magnification of microstructures in Fig. 19. 300X. OLM

