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## Heat Treatment of Duplex Stainless Steel Weld Metals

*High-temperature annealing prevented sigma embrittlement in Ni-enriched welds, but contributed to reduced yield strength*

BY D. J. KOTECKI

**ABSTRACT.** Weld metals for duplex stainless steel Alloys 2205 and Ferralium<sup>®</sup> 255<sup>(1)</sup> were investigated for their response to annealing heat treatments. Those weld metals enriched in nickel for good as-welded properties were found to be embrittled by formation of sigma phase when annealed at 1900°F (1038°C). There was evidence that the sigma forms during heating to 1900°F and some of it dissolves only very slowly at 1900°F. Extended anneals of these weld metals at 1900°F appear to reach an equilibrium in which nearly half of the original ferrite becomes sigma. Annealing at temperatures of 2000°F (1093°C) or higher dissolved the sigma without difficulty. These weld metals could also be annealed above 2000°F, furnace cooled to 1900°F and held for a time, then quenched. Either the higher temperature anneal or the step-anneal resulted in high ductility and toughness but

reduced yield strength. The weld metals with nickel matching the base metals could be annealed at 1900°F without producing sigma, and had adequate ductility but low toughness. In all cases, the annealed yield strength was considerably less than the as-welded yield strength. Higher nitrogen, up to 0.29%, tended to raise the annealed yield strength, but the 80 ksi (550 MPa) requirement for wrought Ferralium 255 could not be consistently achieved.

### Introduction

Duplex austenitic-ferritic stainless steel base metals consist normally of approximately equal percentages of austenite and ferrite. In addition to general corrosion

resistance at least equal to common austenitic stainless, duplex stainless steels have far superior stress corrosion cracking resistance and on the order of double the yield strength as compared to common austenitic stainless. Both the high strength and the stress corrosion cracking resistance of duplex stainless steels are attributable to the mixed microstructure, with usually an assist from alloying with nitrogen (often 0.10 to 0.30% N).

Wrought and cast duplex stainless steels solidify in ingot or casting form as essentially 100% ferrite. Some of that ferrite transforms to austenite during cooling from solidification. Further formation of austenite is achieved by annealing heat treatment, with or without hot working, to produce the final, approximately equal amounts of austenite and ferrite desired for best properties.

Welding of duplex stainless steels presents certain special problems. For weldments to be used in the as-welded condition, filler metal that matches the base metal, because of rapid cooling rates associated with welding, produces a microstructure that is predominantly ferrite. The resulting weldment has poor ductility, poor toughness and relatively poor corrosion resistance. Blumfield, Clark and Guha (Ref. 1), Bryan and Poznanski (Ref. 2), and Kotecki (Ref. 3) have clearly dem-

### KEY WORDS

Duplex Stainless  
Stainless Alloy 2205  
Ferralium 255  
Sigma Embrittlement  
Heat Treatment  
Annealing Treatment  
Sigma Stability  
Flux Cored Weld Wire  
Ni-Enriched Weld Metal  
Step-Annealing

(1) Ferralium 255 is a registered trademark of Haynes International, Kokomo, Ind.

D. J. KOTECKI is with The Lincoln Electric Co., Cleveland, Ohio.

Paper presented at the 69th Annual AWS Meeting, held April 17-22, 1988, in New Orleans, La.

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nance and dropped immediately into a barrel of cold water for quenching.

After heat treatment, a standard all-weld-metal, 1/2-in. (12.5-mm) diameter tensile specimen (Ref. 10) and five Charpy V-notch specimens were machined. The Charpy specimens were oriented transverse to the weld centerline and notched perpendicular to the original plate surface. The tensile specimen was broken at room temperature, and the five Charpy V-notch specimens were broken at -50°F. Charpy results were then averaged for reporting purposes.

After each tensile specimen was fractured, a flat area was ground—finishing with 600-grit abrasive—on one shoulder outside the reduced section parallel to the original plate surface. Five ferrite measurements were taken in this area using a Magne Gage and the procedure given in Ref. 11. The average ferrite number was then calculated.

In addition to the all-weld-metal test plates, an eight-layer all-weld-metal pad for chemical analysis was produced with each experimental wire using the same welding conditions. A complete analysis, including nitrogen, was run on each pad. Since nitrogen can depend to a certain extent on welding conditions with self-shielded flux cored wires (Ref. 12), a second nitrogen check was made on broken tensile specimens, which confirmed the nitrogen analysis of the matching eight-layer pad. Nitrogen results reported herein are those obtained from annealed sigma-free tensile specimens.

When sigma phase was observed in the first welds annealed at 1900°F, it was considered that there could be two possible explanations. The more obvious was that sigma in these weld metals is stable at

1900°F. The second possibility is that sigma was forming during heating to 1900°F and slow to dissolve at 1900°F. When a test plate was placed in the furnace, it brought the furnace temperature down quickly and, as noted earlier, four to five hours was needed to reach 1900°F. So the test plate had to spend considerable time at temperatures where sigma forms rapidly. To test which was really the case, a piece about 1/2 in. long was cut from the shoulder of one tensile specimen that had been annealed at 1900°F (embrittled by sigma), and one that had been annealed at 2000°F (1093°C) or higher (no sigma). This was done with samples from Alloy 2205 weld metal and from Ferralium 255 weld metal. These small samples could be placed directly in the 1900°F furnace and reach temperature very quickly, so they saw very little additional time in the sigma temperature range. Then by withdrawing the samples and quenching after various holding times, one could follow the progress of sigma formation or dissolution by measuring ferrite on the samples. After ferrite measurement, the samples were returned to the 1900°F furnace for further holding at temperature. This was repeated until no further ferrite change occurred. Finally, to be sure that interrupting the anneal at various times was not influencing the results, a second sample from each tensile specimen was annealed at 1900°F for 96 h continuously before quenching.

The manufacturers of Ferralium 255 strongly recommend annealing of this alloy at 1900°F to optimize the partitioning of nitrogen between ferrite and austenite (Ref. 13). This is claimed to optimize corrosion resistance. When sigma was observed in welds annealed 2 to 4 h at

1900°F, longer anneals (up to 16 h) and step-anneals (holding first at 2100°F / 1149°C to be certain that all sigma was dissolved, furnace cooling to 1900°F, and then holding at 1900°F before quench) were considered as alternatives to comply with this recommendation.

### Experimental Results—Alloy 2205 Type Weld Metals

Only two compositions of Alloy 2205-type weld metal were considered. One contained about 8.3% Ni and the other was a matching composition, 6.1% Ni. Table 2 details the test results obtained, both as-welded and heat treated. The 8.3% Ni composition is the same as was developed in the as-welded study (Ref. 3) and shows excellent as-welded properties. The 6.1% Ni weld has surprisingly good ductility in the as-welded condition, in view of the findings of the as-welded study. However, the as-welded toughness of the 6.1% Ni weld would not be acceptable for a 20 ft-lb at -50°F (27) at -46°C) requirement.

The properties of the 8.3% Ni weld annealed at 1900°F (1038°C), as noted in "Reasons for the Present Study" above, were a surprise—very low ductility and toughness. A ferrite measurement on the shoulder of the broken tensile specimen suggested extensive sigma phase, since ferrite transforms to sigma, as noted earlier. Metallographic examination confirmed the presence of extensive sigma, which was not present in the as-welded condition—Fig. 1. In contrast, the 8.3% Ni weld annealed at 2000°F (1093°C) was free of any traces of sigma—Fig. 2. The etchant, Kalling's reagent, darkens ferrite rapidly, attacks sigma more slowly, and

Table 2—Alloy 2205-Type Weld Metal Results

Experimental Wire		9292-109				9292-204			
Deposit C		0.028				0.027			
Mn		1.77				0.67			
P		0.025				0.007			
S		0.006				0.011			
Si		0.46				0.26			
Cr		21.60				22.61			
Ni		8.32				6.11			
Mo		3.15				3.10			
N		0.12				0.193			
Anneal, h	As-	2	4	4	4	As-	4	4	2
@ Temp., F°(C°)	Welded	1900	1950	2000	2050	Welded	1900	2100	1900
		(1038)	(1066)	(1093)	(1121)		(1038)	(1149)	(1038)
Tensile, ksi	117.0	109.4	106.0	103.0	105.0	108.4	108.0	104.0	
(MPa)	(807)	(754)	(731)	(710)	(724)	(747)	(745)	(717)	
Yield, ksi	86.0	72.5	65.0	66.0	66.2	85.3	66.8	64.9	
(MPa)	(593)	(500)	(448)	(455)	(456)	(588)	(461)	(447)	
% Elongation	30.5	4.5	28	36	36.0	28	33.5	39.0	
Charpy V@	38.0	2.3	49	52	53.6	18.5	17.1	16.2	
-50°F/-46°C,	(51.5)	(3.1)	(66.4)	(70.5)	(72.7)	(25.1)	(23.2)	(22.0)	
ft-lb (J)									
Ferrite of	46	4	20	43	42	88	61	58	
Tensile Specimen,									
FN									



Table 3—Ferrallium 255 Type Weld Metal

Experiment	9292-112	9292-136	9292-137	9292-161	9292-174	9292-202	9292-203	9292-231	9292-358	9292-622	9292-650	9292-678
C	0.014	0.019	0.020	0.022	0.025	0.024	0.023	0.018	0.019	0.023	0.019	0.018
Mn	1.06	1.11	1.08	0.94	0.97	0.59	0.92	0.95	1.39	1.38	1.36	1.33
P	0.026	0.033	0.028	0.029	0.028	0.004	0.030	0.033	0.029	0.025	0.033	0.026
S	0.005	0.010	0.009	0.005	0.005	0.011	0.007	0.004	0.006	0.015	0.008	0.009
Si	0.36	0.61	0.78	0.33	0.25	0.03	0.20	0.37	0.17	0.22	0.37	0.16
Cr	24.51	23.41	23.31	24.26	24.34	25.20	25.08	25.12	26.26	25.31	25.39	26.29
Ni	10.17	7.31	7.18	8.96	8.46	5.85	9.04	9.55	9.20	9.35	9.42	9.52
Mo	3.18	3.29	3.77	2.93	2.74	2.99	3.01	3.41	3.40	3.30	2.93	3.40
Cu	2.14	2.00	2.06	1.70	1.93	2.04	1.97	2.00	1.90	2.14	2.09	2.13
N	0.100	0.203	0.217	0.154	0.194	0.173	0.152	0.134	0.249	0.256	0.266	0.289
As-Welded Properties												
Tensile, ksi (MPa)	121.8 (840)	127.6 (880)	131.8 (909)	120.1 (828)	123.2 (849)	120.8 (833)	125.2 (863)	126.8 (874)	131.3 (905)	120.2 (829)	123.5 (852)	132.3 (912)
Yield, ksi (MPa)	89.6 (618)	98.9 (682)	105.0 (724)	93.1 (642)	89.9 (620)	98.6 (680)	98.0 (676)	100.1 (690)	98.9 (682)	89.4 (616)	94.2 (650)	99.8 (688)
% Elong. in 2 in.	24.5	21.5	18.0	28.0	31.0	22.5	26.0	13.0	27.5	30.0	33.0	30.5
ft-lb (J) @ -50°F	26.3 (35.7)	13.3 (18.0)	13.2 (17.9)	30.4 (41.2)	22.1 (30.0)	10.6 (14.4)	24.1 (32.7)	23.6 (32.0)	18.2 (24.7)	18.0 (24.4)	15.1 (20.5)	15.1 (20.5)
Ferrite, FN	45	42	53	64	50	100	54	58	53	34	42	36
Properties After Anneal With at Least Some Embrittlement												
h @ °F	2 @ 1900	—	—	4 @ 1900	4 @ 1900	—	4 @ 1900	2 @ 1900	4 @ 1900	4 @ 1900	4 @ 1900	4 @ 1900
Tensile, ksi (MPa)	106.1 (732)	—	—	110.5 (762)	116.2 (801)	—	114.7 (791)	101.1 (697)	122.0 (841)	123.2 (849)	114.1 (787)	110.2 (760)
Yield, ksi (MPa)	79.1 (545)	—	—	65.7 (453)	73.4 (506)	—	69.3 (478)	85.1 (587)	92.9 (641)	76.4 (527)	81.5 (562)	92.6 (638)
% Elong. in 2 in.	1.0	—	—	37.0	34.0	—	7.0	1.5	2.0	10.0	2.0	1.0
ft-lb (J) @ -50°F	2.6 (3.5)	—	—	49.5 (67.1)	10.3 (14.0)	—	2.3 (3.1)	2.2 (3.0)	1.4 (1.9)	2.2 (3.0)	1.9 (2.6)	1.4 (1.9)
Ferrite, FN	2	—	—	40	39	—	28	1	5	16	4	3
Properties After Anneal With No Embrittlement												
hr @ °F	2 @ 2050	4 @ 2000	4 @ 2000	4 @ 2100	4 @ 2050	4 @ 1900	—	2 @ 2100	4 @ 2100	4 @ 2100	4 @ 2100	4 @ 2100
Tensile, ksi (MPa)	106.2 (732)	118.5 (817)	120.7 (832)	109.8 (757)	114.4 (789)	114.7 (791)	—	117.1 (807)	125.7 (867)	117.6 (811)	117.1 (807)	124.3 (857)
Yield, ksi (MPa)	65.7 (453)	73.0 (503)	73.6 (507)	67.0 (462)	72.2 (498)	71.4 (492)	—	73.8 (509)	80.0 (552)	72.3 (499)	73.8 (509)	79.2 (546)
% Elong. in 2 in.	33.0	32.0	33.5	33.5	36.0	30.0	—	32.0	30.0	35.5	39.5	38.0
ft-lb (J) @ -50°F	52.1 (70.6)	21.4 (29.0)	23.9 (32.4)	53.5 (72.5)	32.7 (44.3)	12.0 (16.3)	—	52.6 (71.3)	26.6 (36.1)	32.8 (44.5)	30.2 (41.0)	25.0 (33.9)
Ferrite, FN	45	44	48	43	39	66	—	52	44	30	28	41
Properties After Step Anneal With No Embrittlement												
h @ 2100 °F	2	—	—	2	2	4	4	—	—	—	—	4
h @ 1900 °F	2	—	—	2	2	2	2	—	—	—	—	2
Tensile, ksi (MPa)	107.3 (740)	—	—	108.8 (750)	112.4 (775)	112.9 (778)	113.7 (784)	—	—	—	—	126.3 (871)
Yield, ksi (MPa)	60.4 (416)	—	—	63.4 (437)	67.4 (465)	72.8 (502)	65.7 (453)	—	—	—	—	76.2 (525)
% Elong. in 2 in.	34.0	—	—	34.0	34.5	34.0	35.0	—	—	—	—	29.0
ft-lb (J) @ -50°F	50.6 (68.6)	—	—	25.4 (34.4)	32.3 (43.8)	9.5 (12.9)	37.5 (50.9)	—	—	—	—	21.0 (28.5)
Ferrite, FN	42	—	—	51	46	66	45	—	—	—	—	35

Temperature conversions: -50°F = -46°C, 1900°F = 1038°C, 2000°F = 1093°C, 2050°F = 1121°C, 2100°F = 1149°C.







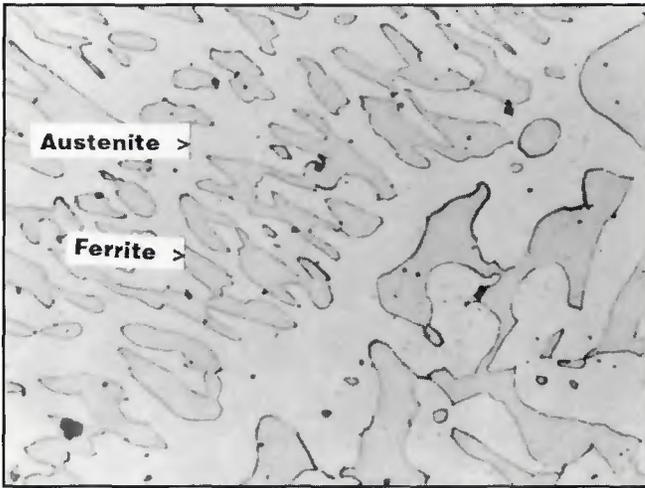


Fig. 15—Microstructure of weld from Experiment 9292-678 (26% Cr, 0.29% N Ferralium 255). Annealed 4 h at 2100°F (1149°C)

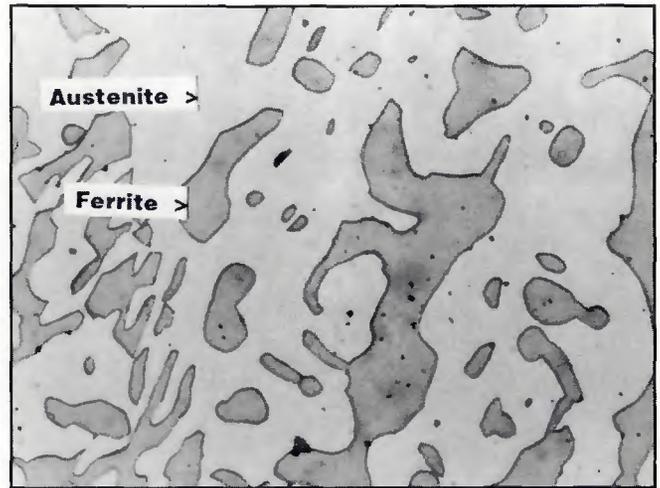


Fig. 16—Microstructure of weld from Experiment 9292-678 (26% Cr, 0.29% N Ferralium 255). Step-annealed 4 h at 2100°F (1149°C) and 2 h at 1900°F (1038°C)

somewhat lower yield strength than the simple 2100°F anneal.

Figures 15 and 16 compare the microstructures resulting from simple 2100°F anneal and from step-anneal of Experiment 9292-678. It will be noted that a different etchant was used for these samples as compared to previous microstructures. Kalling's Reagent had no effect on this higher chromium weld metal (a micro was not prepared of the other 26% Cr weld, Experiment 9292-358) at room temperature. Heating the Kalling's Reagent caused severe pitting in the ferrite. Murakami's Reagent was tried at room temperature also, with no effect. Heating Murakami's Reagent slowly with the specimen immersed produced the desired etching with only very minor pitting. So, these microstructures are presented after etching with hot Murakami's Reagent. No sigma is present either in the 2100°F annealed weld metal nor in the step-annealed weld metal. The step-annealed microstructure may be somewhat coarser.

## Discussion of Results

This study has considered three main concerns relating to annealed duplex stainless steel weld metals. These are:

- 1) the formation and stability of sigma phase during annealing,
- 2) the achievement of base metal yield strength requirements in sigma-free weld metal,
- 3) the achievement of a supplementary impact requirement of 20 ft-lb at -50°F (27) at -46°C.

Most of the compositions considered herein were embrittled by sigma phase after 1900°F (1038°C) anneals. Both the low-nickel weld metals that matched their respective base metals, 9292-204 (Alloy 2205, Table 2) and 9292-902 (Ferralium 255, Table 3), were exceptions. It is quite

clear from this that higher nickel in the weld metal increases the stability of sigma. Weld metals of increased nickel for good as-welded properties have to be annealed at higher temperatures than weld metals that match base metal composition in order to avoid sigma.

It appears that high chromium and high molybdenum also work to stabilize sigma. Of course, one could aim weld metal compositions at the bottom of the acceptable ranges, but this would be at the risk of inferior corrosion resistance in the weld metal. High chromium, and especially high molybdenum, are preferred for pitting resistance in particular, which is why most compositions examined herein were above 3% Mo. High nitrogen, up to nearly 0.3%, did not appear to be very effective in preventing sigma formation.

If a final annealing temperature of 1900°F must be used for optimum corrosion resistance, as suggested by Ref. 13, with high-nickel weld metal, then step-annealing seems a viable method of achieving this. It must be accepted that any sizeable weldment cannot be heated fast enough to annealing temperatures to prevent sigma formation during heating. If one overshoots the final annealing temperature to put all sigma back into solution, then when the weldment cools back to 1900°F, it becomes necessary to nucleate sigma again (assuming that a high-nickel composition is in use so that sigma will be stable at 1900°F). This nucleation apparently takes some time. Ferralium 255-type weld metals high in nickel, which formed vast quantities of sigma in a simple 1900°F anneal, were sigma-free when step-annealed, first at 2100°F (1149°C), then furnace cooled to 1900°F and held for 2 h before quenching. So it must take more than 2 h to nucleate sigma at 1900°F, provided that the weldment can get to 1900°F sigma-free. It can do this

from higher temperatures, but not from lower temperatures.

In order to obtain higher weld metal yield strengths after sigma-free anneals, higher nitrogen, especially in combination with higher chromium, seemed beneficial in this study. Figure 17 presents the sigma-free anneal data of Table 3 (not step-annealed) as a plot of yield strength versus nitrogen content. The nitrogen values are those measured in these broken tensile specimens. A regression line is included in Fig. 17. The correlation coefficient,  $R^2$ , is not large, but the trend of increasing annealed yield strength with increasing nitrogen content is clear.

Coupled with high nitrogen, high chromium seems beneficial for obtaining high annealed yield strength. The two highest annealed yield strengths in Ferralium 255-type weld metal were achieved with over 26% Cr, Experiments 9292-358 and 9292-678—Table 3. Perhaps the higher chromium makes nitrogen more soluble in ferrite. At any rate, it enhances the overall solubility of nitrogen in the weld metal—porosity was not observed even at 0.29% nitrogen with 26% Cr. When the weld metal solidifies, it is entirely ferrite, so the solubility of nitrogen in ferrite at the solidification temperature, at least, must be high to avert porosity.

It is unsettling that, despite nearly 0.3% N and over 26% Cr, the 80 ksi (550 MPa) yield strength requirement of Ferralium 255 base metal was not comfortably exceeded, in the annealed condition. The same weld metals that, in the as-welded condition, reached close to 100 ksi (695 MPa), lost on the order of 20% of their as-welded yield strength in annealing, without losing much ferrite, if any.

If ferrite is not being lost due to annealing, then one must wonder what is the mechanism whereby the yield strength drops in annealing. Some apparent micro-



