

The Dislocation Density of Acicular Ferrite in Steel Welds

It is estimated that the dislocation density of acicular ferrite contributes 21 ksi to its strength

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ABSTRACT. The dislocation density of acicular ferrite in a steel weld deposit has been estimated using transmission electron microscopy, at about 10^{14} m^{-2} , contributing approximately 145 MPa (21 ksi) to its strength.

Introduction

Acicular ferrite is a phase formed by the transformation of austenite during cooling of low-alloy steel weld deposits (Ref. 1). It exhibits a thin-plate morphology and forms in a temperature range where reconstructive transformations become relatively sluggish and give way to displacing transformations. The transformation is found to exhibit an "incomplete-reaction phenomenon" in the sense that the formation of acicular ferrite ceases before the residual austenite reaches its equilibrium composition (Refs. 2, 3). The growth of acicular ferrite is known to be accompanied by an invariant-plane strain shape deformation (Ref. 3). Since the transformation occurs at fairly high temperatures where the yield strengths of the phases concerned are relatively low, the shape change may to some extent be plastically accommodated. This plastic deformation would in turn cause the dislocation density of the acicular ferrite and any residual austenite to increase. While it is known qualitatively that the dislocation density of acicular ferrite is fairly high (Refs. 4, 5), there are few quantitative data to this effect. A recent review (Ref. 6) quoted the dislocation density to be $10^{12} - 10^{14} \text{ m}^{-2}$, based on the work of Tuliani (Ref. 7) and Watson (Ref. 8), although the details of the measurements were not mentioned. The work presented here is part of a program of research on the quantitative pre-

diction of weld metal microstructure and properties. It deals specifically with the measurement of the dislocation density of acicular ferrite, with a view to estimating the contribution of dislocations to the strength of acicular ferrite.

Experimental Techniques

The specimens studied were taken from the top layer of a manual metal arc weld of chemical composition: Fe-0.031C-0.40Si-1.68Mn-2.46Ni-0.17Mo wt-% (the weld also contained the following elements: 0.04Cr-0.01V-0.005S-0.008P-0.02Al-0.03Ti-0.01Nb-0.0333O-0.0080N wt-%). The chemical analysis was carried out using a spectroscopic technique, although the concentrations of oxygen and nitrogen were measured using *Leco* furnaces (Ro-17 and Tn-15), with 50 g of material for each determination to ensure representative results. The welds were made using 4-mm ($5/32$ -in.) diameter electrodes (E10016-G type, as defined by the American Welding Society); the joint geometry was designed according to BS639 in order to avoid dilution from the base plate. Welding was carried out in the flat position, using the stringer bead technique; the base plate thickness was 20 mm (0.8 in.). The welding current and voltage used were 180 A and 23 V (DCEP), respectively (nominal arc energy $\approx 2 \text{ kJ mm}^{-1}$); the weld consisted of some 21 passes with 3 passes per layer, deposited at a speed of

about 0.002 m s^{-1} . The interpass temperature was typically 250°C (482°F).

Transmission electron microscopy samples were prepared from 3-mm (0.12-in.) diameter disks machined from the top layer of the weld, containing the as-deposited, primary microstructure. The disks were mechanically ground down to a thickness of 0.08 mm (0.003 in.) on 1200-grit SiC paper. The specimens were then twin-jet electropolished using a 5% perchloric acid, 25% glycerol and 70% ethanol mixture at ambient temperature and 45 V. They were examined using a Phillips EM400T transmission electron microscope operated at 120 kV.

Results and Discussion

The dislocation structure observed in thin foil samples can be approximately representative of the bulk material if precautions are taken during the preparation of foils. It is unlikely that dislocations are introduced during thinning, but the subsequent handling of specimens can lead to accidental deformation. The dislocations introduced in this way tend to be long and nearly straight (Ref. 9) since they lie parallel to the foil surface. This damage is easily recognized with experience and can be avoided in ordinary polycrystalline specimens. In any event, great care was exercised during the handling of thin foil specimens. Other factors such as surface image forces may make the dislocation leave the foil, but the use of oxidizing polishing solutions usually leaves a thin oxide film on the surface of the foil and prevents any such losses.

Each dislocation density determination was based on measurements from ten micrographs, taken at magnifications of 60,000 to 100,000. These magnifications were chosen because they are high enough to resolve individual dislocations; at the same time, the magnifications are low enough to ensure that the micrographs had the same apparent dislocation content as the surrounding regions. Ham (Ref. 10) and Hirsch, *et al.* (Ref. 11),

KEY WORDS

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