

Fig. 1—Microstructure of AA5754 base material used in the experiment.

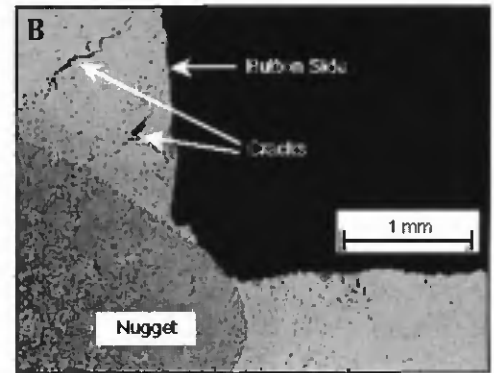
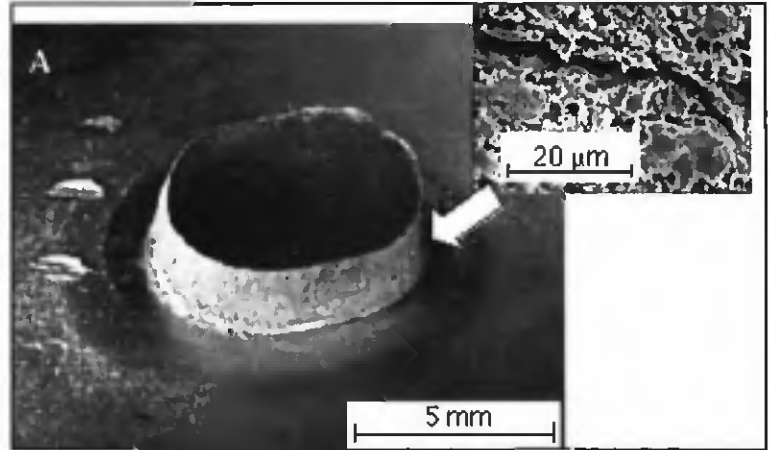


Fig. 2—Appearance of a typical button after peel testing. A — An amplified side view of the button wall; B — cross section of the same button.

It is noteworthy to compare cracking during welding with solidification cracking during casting, although there are differences in the processes. According to the classical works by Pellini and Flemings, hot tearing in casting alloys occurs at the last stage of crystallization (Refs. 18–21), during which solid grains are surrounded by the liquid; such a structure has a very low strength. Tensile stresses and strains, resulting from nonuniform temperature distribution and cooling, may cause material failure. A certain amount of strain is necessary for crack initiation, as pointed out by Pellini (Ref. 19). Hot cracking tendencies in casting increase with grain dimensions, solidus-liquidus gap and solidification shrinkage, which is especially high for Al alloys. The presence of impurities and grain boundary segregation also promote cracking.

The mechanism of hot cracking in welding, similar to that in casting, is based on a theory developed by Borland (Ref. 22) and Prokhorov (Ref. 23). Occurrence of cracking in "coherence temperature range" (Borland's definition) depends on both critical strain and critical strain rate. Hot cracking during welding at elevated, near-solidus temperatures includes failure of welds (solidification cracking) and cracking in the HAZ (liquation cracking) (Ref. 24). Cracking in the HAZ is related to liquation at the grain boundaries of either the secondary phase or low-melting-point impurities, at subsolidus and at supersolidus temperatures of the primary phase. Existing theories of formation and solidification of grain boundary liquid films include equilibrium melting of the vicinity of grain boundaries (Ref. 25), constitutional liquation of secondary phases and the effects of segregation (Refs. 26, 27).

Comparisons of various Al alloys in casting and arc welding revealed the Al-Mg system is second to the Al-Cu system in crack susceptibility among aluminum alloys (Refs. 16, 18, 21, 28), in spite of only a small amount of eutectic formed

during solidification (Ref. 16).

The lack of practical information on cracking in spot welded Al-Mg alloys, the increasing use of Al alloys in the automotive industry and the care needed for spot welding aluminum alloys are the driving forces for the study of cracking in resistance spot welding AA5754 sheets. Because single spot welds are rarely used in welded structures, multispot welds were chosen for this investigation. The influence of metallurgical interaction among spot welds and other factors are emphasized to understand the mechanisms of crack initiation and propagation.

Experiments

AA5754 aluminum alloy sheets of 1.6- and 2.0-mm gauges and in 0 temper condition (annealed), produced by Alcan Aluminum Co., were used in all experiments. The sheet surface was pretreated and prelubricated by the producer. The chemical composition specified by the producer, as well as the composition of the sample tested independently, are listed in Table 1. The data show that the tested composition is within the specified range.

The sheets were cut into 350 x 25-mm coupons for multiwelding. Taking into account the small anisotropy of structure and properties reported by Burger, *et al.* (Ref. 5), all coupons were cut out parallel to the rolling direction. Figure 1 shows a typical material microstructure of the base metal. Slightly elongated grains representing Mg in Al solid solution are visible as are precipitates of Al_3Mg_2 , (Fe, Mn) Al_6 and silicides.

In this set of experiments, Alcan's domed electrodes were used. Such an electrode has a face diameter of 10 mm, and the radius for the domed face is 50 mm.

Multiwelds were made on the coupons using a medium-frequency (MF) DC welding machine. The welding para-

meters used were 7 kN electrode force, 3 kA preheat current for 3 cycles (50 ms), 12 cycles (200 ms) of weld delay, 26 kA welding current for 5 cycles (83 ms) and 12 cycles (200 ms) of holding time. The weld pitch was 30 mm. Generally, welds with satisfactory appearance were obtained. Commonly used peel testing confirmed good quality and repeatability of the spot welding process. A regularly shaped, good sized button is shown in Fig. 2A.

Spot welded samples were then sectioned in two perpendicular directions (normal and parallel to the rolling direction) and were ground, polished and structurally investigated to disclose possible internal discontinuities and their natures. Optical and scanning electron microscopy techniques were used, as were energy dispersive X-ray (EDX) and wave dispersive X-ray (WDX) microanalyses. Microhardness was measured across the sectioned weldment using a LECO hardness tester.

Description of Cracks

Despite the normal appearance of a typical button after peeling, an amplified side view of the button wall revealed

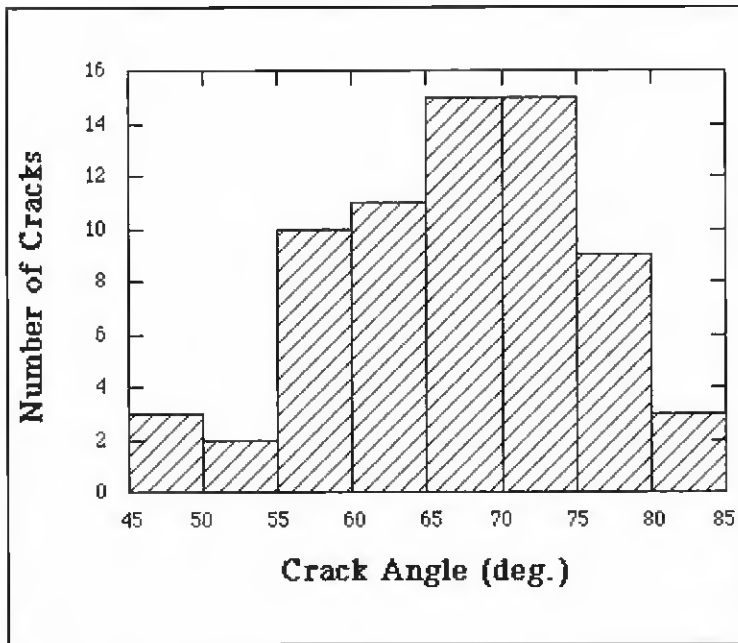


Fig. 4 — Measured crack angles with respect to weld interface.

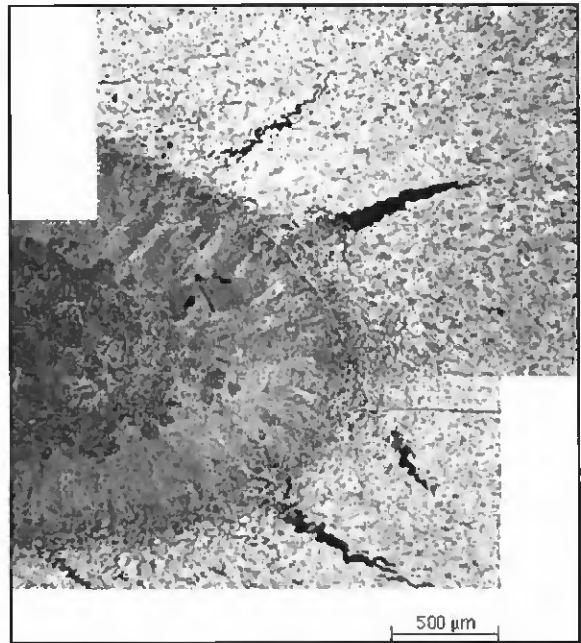


Fig. 5 — A close look at cracks in the HAZ.

and, therefore, it is liquation cracking according to the classification by Hemsworth, *et al.* (Ref. 24). The micro-porosity visible in some inclusions of the secondary phase serves as additional evidence of the existence of liquid in this part of the HAZ during spot welding.

Generally, there are two possible ways for melting to occur at grain boundaries in the HAZ: at supersolidus temperatures and at subsolidus temperatures. In the heating stage of welding, equilibrium melting of the material near grain boundaries occurs in the part of the HAZ heated to the temperature range between solidus and liquidus (partially melted zone). In addition to partially melting at temperatures above the solidus, liquation of the secondary phase may occur at subsolidus temperatures. During rapid heating, which is a characteristic of resistance spot welding, there is not sufficient time to dissolve the Al_3Mg_2 phase in the α -solution matrix, and inclusions of this phase still exist after the alloy is heated over the solvus line. Al_3Mg_2 inclusions melt in the region (next to the partially melting zone) that experiences maximum temperature above the eutectic point but below the solidus temperature. Existence of liquid below the solidus of AA5754 can also be attributed to other low-temperature melting additions/impurities present in a commercial alloy. The zones around the nugget are schematically shown in Fig. 10. Structures/zones in the HAZ are linked to the equilibrium phase diagram via assumed temperature history during RSW. The dy-

namic effects in the heating/cooling processes, such as overheating and undercooling, were not investigated in this study due to experimental difficulties, although they may contribute to hot cracking. For instance, the effective solidus temperature during cooling may be lower than the equilibrium solidus temperature because of the high cooling rate in RSW. This effectively enlarges the temperature range in which the material is weak and susceptible to cracking.

As a result of combined supersolidus and subsolidus melting/liquation, large grains in these parts of the HAZ are surrounded by liquid during welding. Nearly continuous films of liquid are formed at grain boundaries — Fig. 8. Therefore, the overall material structure close to the nugget is favorable for crack initiation and growth during the last stage of heating in resistance spot welding Al-Mg alloys.

After the current is switched off, the material cools quickly because of heat transfer through the water-cooled electrodes. The life of transient liquid films at grain boundaries depends on several factors, including the cooling rate and compositional segregation. For AA5754, the difference between equilibrium temperatures of liquidus (915 K), solidus (876 K) and eutectic temperature (723 K) is significant, according to the Al-Mg phase diagram (Ref. 29). Decreases of the solidus and eutectic solidification temperatures can be expected due to kinetic effects during cooling. The coexistence of solid and liquid phases at grain bound-

aries is then extended in a relatively wide temperature range during the cooling stage.

High heating and cooling rates, as well as a high temperature gradient in the weldment due to the nature of Joule heating, are the thermal characteristics of spot welding. Based on the experiments conducted in this study of welding AA5754, the heating and cooling rates are estimated to be as high as 8000 and 3000 K/s, respectively. Because of this, liquid films at grain boundaries may exist for an extended time at elevated temperatures, as proved by Radhakrishnan and Thompson in their model (Ref. 27). This results from the concentration gradient in the liquid due to rapid solidification, which effectively lowers the solidification temperatures of liquid parts with higher (than equilibrium) Mg concentration.

As described above, metallurgical factors during spot welding of AA5754 create favorable conditions for tearing the structures in the HAZ.

Thermomechanical Factors

Besides the metallurgical effect, thermomechanical factors also play a role in the initiation and subsequent propagation and growth of cracks. This section is devoted to describing the mechanisms of crack formation by qualitative thermal and mechanical analyses using simplified assumptions (because of the complicated mechanical, thermal and metallurgical interactions).

As seen in Fig. 3A, cracks appear on

